CHEMISTRY PROJECTS FOR CLASS 12

(DETERMINATION OF THE CONTENTS OF Soft Drink)-5TH PARTGANESH KUMARDATE:- 24/02/2021

TEST FOR CARBON DIOXIDE

Experiment:

As soon as the bottles were opened one by one the samples were passed through LIME WATER the lime water turns MILKY.

Observation:

SL. No.	Name of drink	Time taken (in sec)	Conclusion
1	Coca Cola	26.3	CO2 present
2	Sprite	21	CO2present
3	Limca	25	CO ₂ present
4	Fanta	36	CO ₂ present

<u>Inference</u>

All soft drinks contain dissolved CO2

Chemical Reaction involved:

ightarrow Ca (OH)₂+CO₂ \Longrightarrow CaCO₃ +H₂O

TEST FOR GLUCOSE

Experiment:

Observation:

Glucose is reducing sugar acid its presence is detected by the following test. Benedict's reagent test:

Small sample of cold drink of different brands were taken in a test tube and a few drops of Benedict's reagent were added to it test tube was heated for a few seconds Reddish coloration confirmed the presence of glucose in the samples.

SL No. Name of drink **Observation Conclusion Reddish color Glucose is present** Coca Cola 1 ppt **Sprite Reddish color Glucose is present** 2 Ppt **Reddish color Glucose is present** 3 Limca ppt **Reddish color Glucose is present** 4 Fanta ppt

<u>Inference</u> All the samples gave positive test for glucose with Benedict's reagent hence all of them contain glucose

TEST FOR ALCOHOL

<u>Experiment :</u>

Small samples of cold drinks of different brands were taken in separate test tube and iodine followed by Potassium iodide and Sodium Hydroxide (NaOH) solution was heated in hot water for 30 minutes. Presence of yellow colored ppt. confirmed the presence of alcohol.

SL No .	Name of drink	Observation	Conclusion
1	Coca Cola	Yellow ppt	Alcohol present
2	Sprite	Yellow ppt	Alcohol present
3	Limca	Yellow ppt	Alcohol present
4	Fanta	Yellow ppt	Alcohol present

Inference:

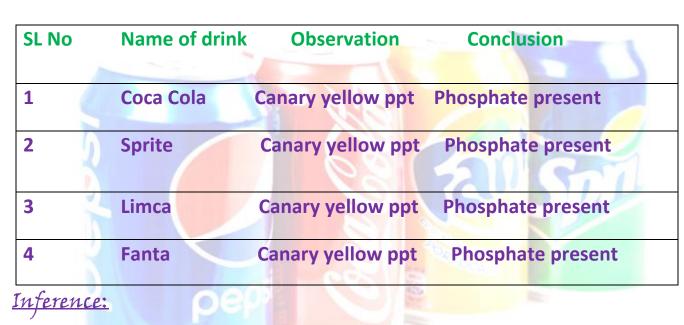
All samples contain as they give positive test for it

TEST FOR PHOSPHATE

<u>Experiment:</u>

Small samples of the drinks of different brands were taken in separate test tubes and ammonium Molybdate followed by nitric acid HNO was added to it Presence of canary yellow ppt confirmed the presence of phosphate ions in cold drinks.

<u>Observation:</u>



All samples contain phosphate as they give positive test for it

Chemical Reactions involved:

 $NaHP0_{4}+12(NH_{4_{2}})MnO_{4}+21HNO_{3}+3H \Longrightarrow$ $(NH_{4})_{3}PO_{4}M_{0}O_{3}+21HN_{4}NO_{7}+12H_{2}O$
